

## Applications of Selected Nano-Particles in Nano-Medicine

Nano-Biotechnology PL 2012  
Malej Auli Politechniki, pl. Polytechniki 1  
Warsaw, Poland  
17-18 September 2012

**Dipl.-Chem. Helmut Schmid**

**Fraunhofer-Institute Chemical Technology (ICT), Box 1240, D-76318 Pfinztal, Germany**  
**Web: [www.ict.fhg.de](http://www.ict.fhg.de), Phone: +49 721 4640-709, Mail: [sd@ict.fhg.de](mailto:sd@ict.fhg.de)**



---

# Index

---

2

## **1 Introduction**

- 1.1 Fraunhofer ICT
- 1.2 Classification (Preferred Particle Sizes)
- 1.3 Technical Applications
- 1.4 Medical Applications

## **2 Methods and General Tools**

- 2.1 Nano-Production, Stabilization
- 2.2 Analysis / Characterization
- 2.3 Processing-Technology

## **3 Results / Discussion, Selected Applications**

## **4 Summary / Conclusion**

## **5 Prospective**

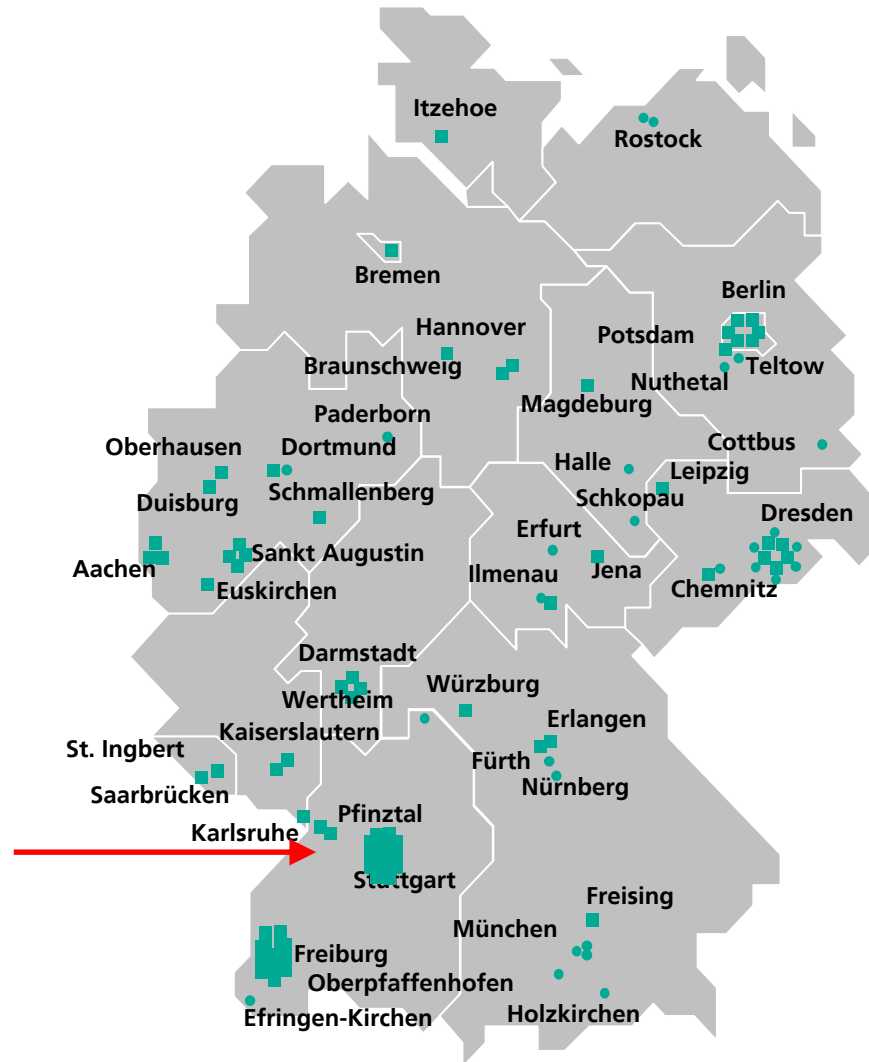


# 1 Introduction

3

## Fraunhofer-Society Locations in Germany

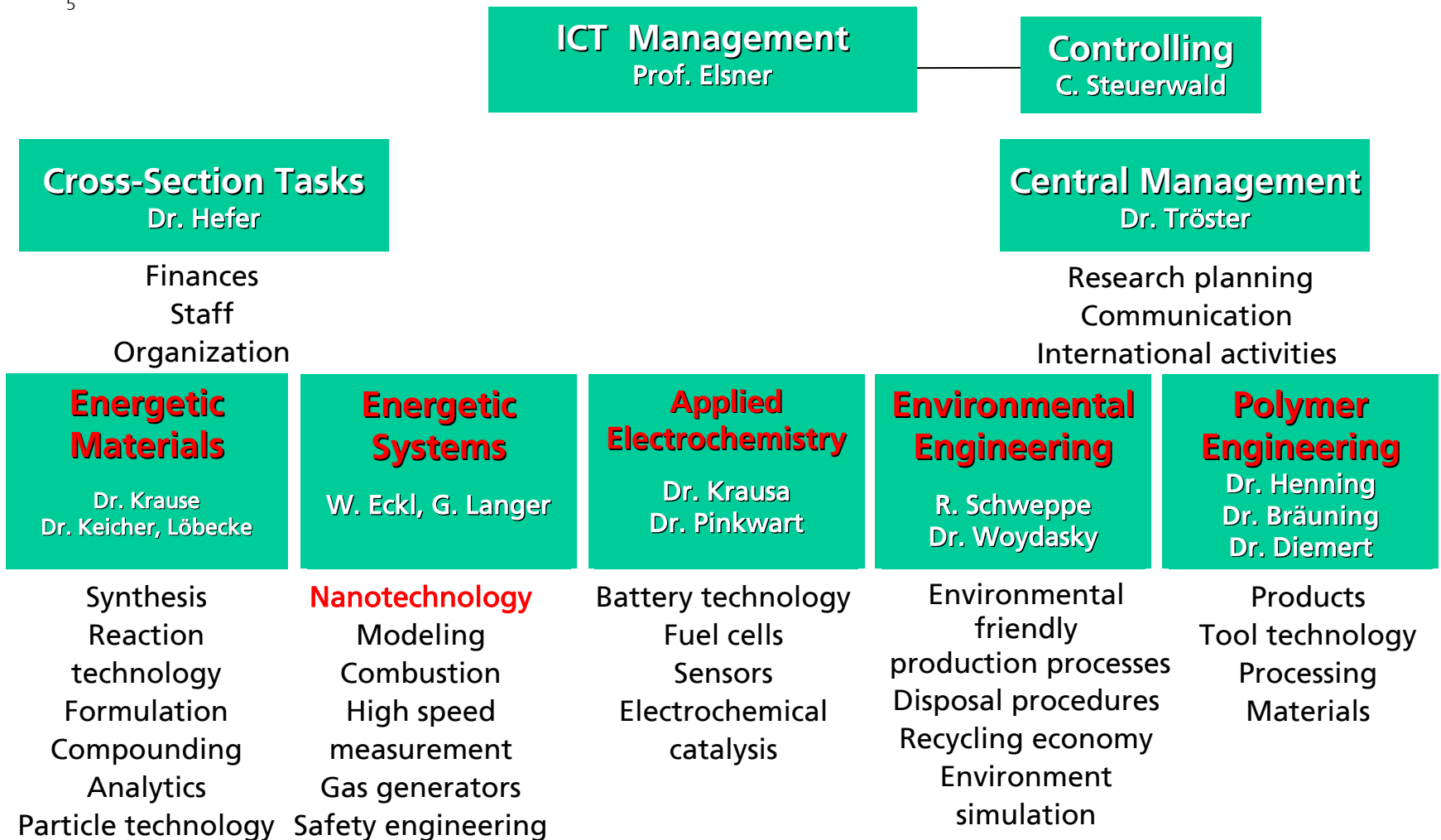
- 56 Institutes
- 12 770 Staff
- More than 40 locations in Germany
- International representations





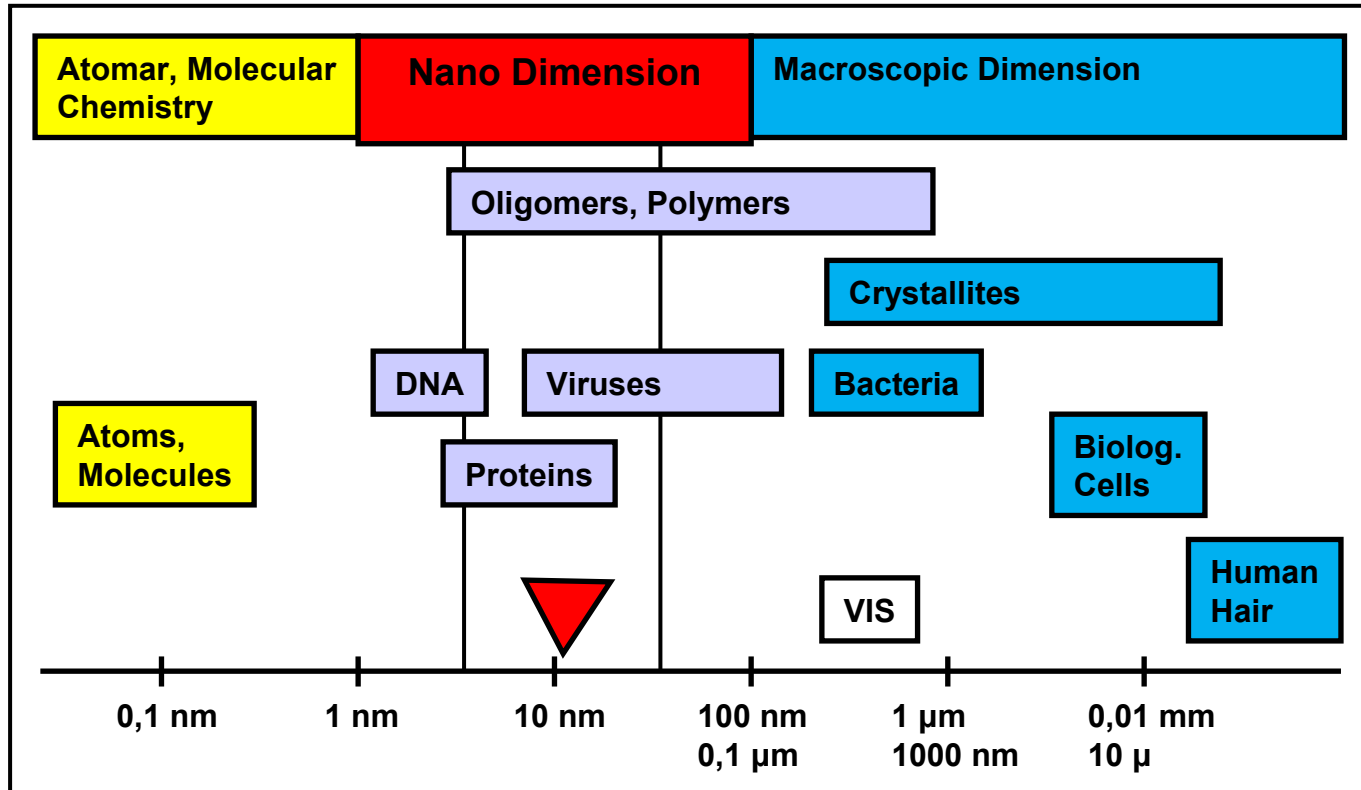






# Definition

6



**Figure 6.1**

Size classification of well known objects. Interesting Nano-Region: 5 – 50 nm.



# Medical Applications

7

## General Medical Research

1. Drug Discovery
2. Analysis of Cell Signal Pathways

## Therapy

1. Drug Delivery (Targeting, Controlled Release)
2. Gene Delivery
3. Personalized Medicine
4. Regenerative Medicine (Bioactive Surfaces)
5. Tissue Engineering
6. Antimicrobial Surfaces
7. Cancer Therapy

## Diagnostics In-vitro, In-vivo

1. Biosensors
2. Biochips (DNA, Proteins, Cells)
3. Medical Imaging (Ultrasonic, NMR, X-Ray)

„Thera-Nostics“

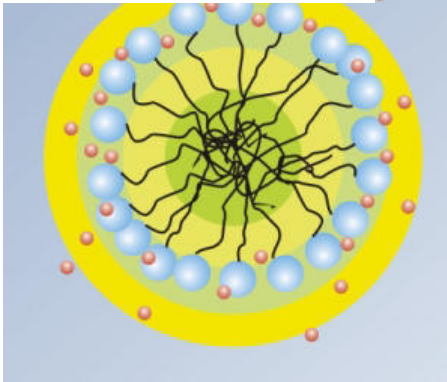


# Organic and Inorganic Nano-Particles as Drugs / Carriers

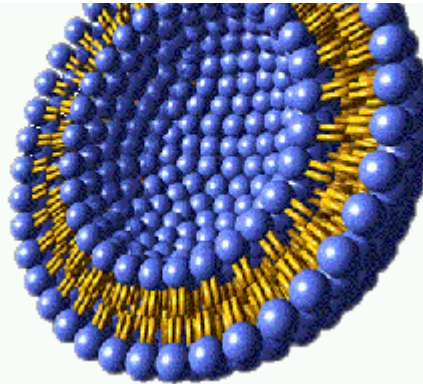
8

Sources: Uni Dortmund, Lenz

## Micells (Vesicles)



## Liposomes (Vesicles), Phospho-Lipids



## Polymer-Protein-Conjugates

### Complex Compounds

Gd-DTPA (Chelate)

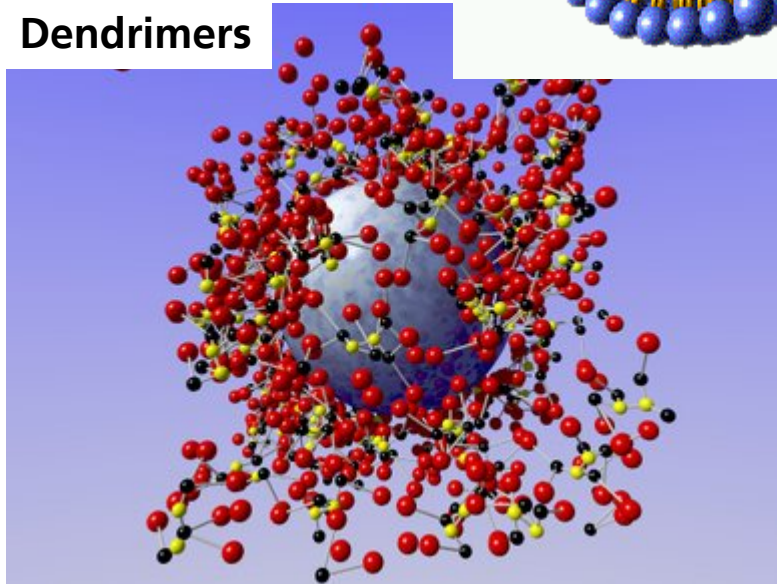
Gadolinium-diethylene-triamino-pentaacetate

$Gd^{3+}$ , 7 unpaired f-e<sup>-</sup>, Paramagnetism

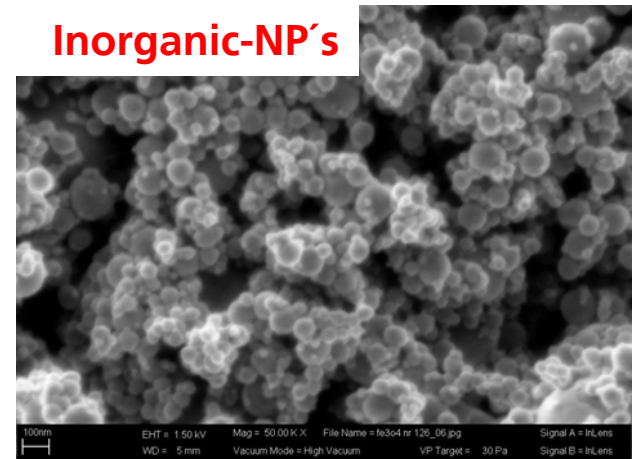
Contrast Agent MRI:

Toxicity -> Nephrogenic Systemic Fibrosis

## Dendrimers



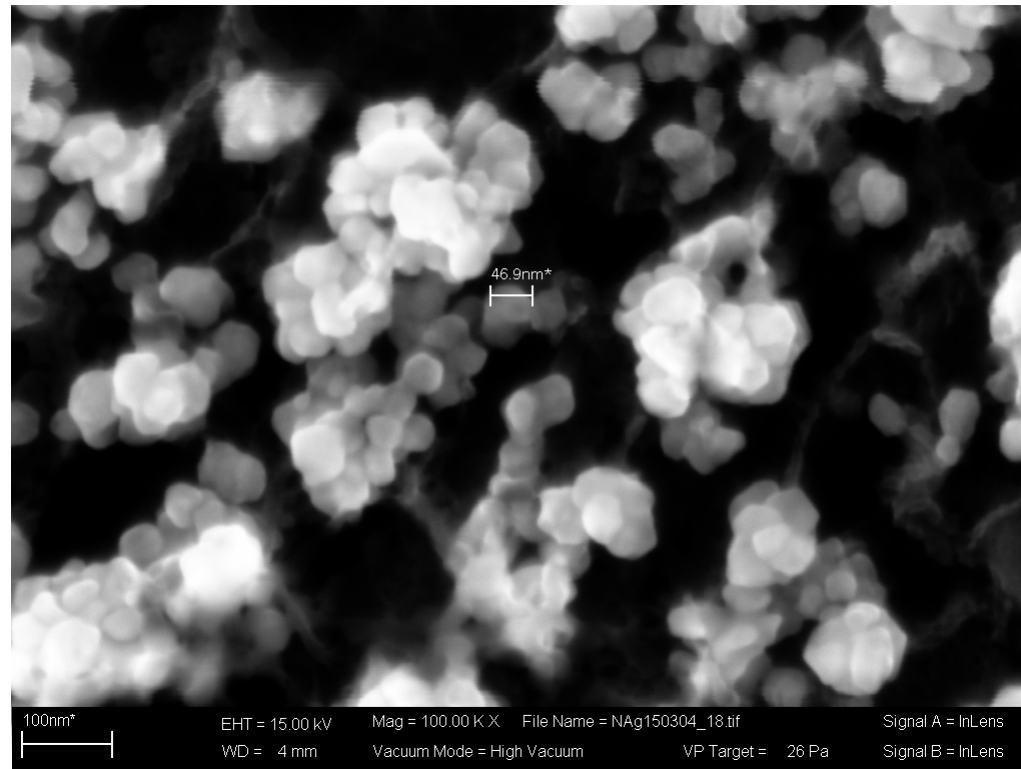
## Inorganic-NP's





## Example: Nano-Silver with Antimicrobial Functionality

9



File: NAg150304\_18.Tif

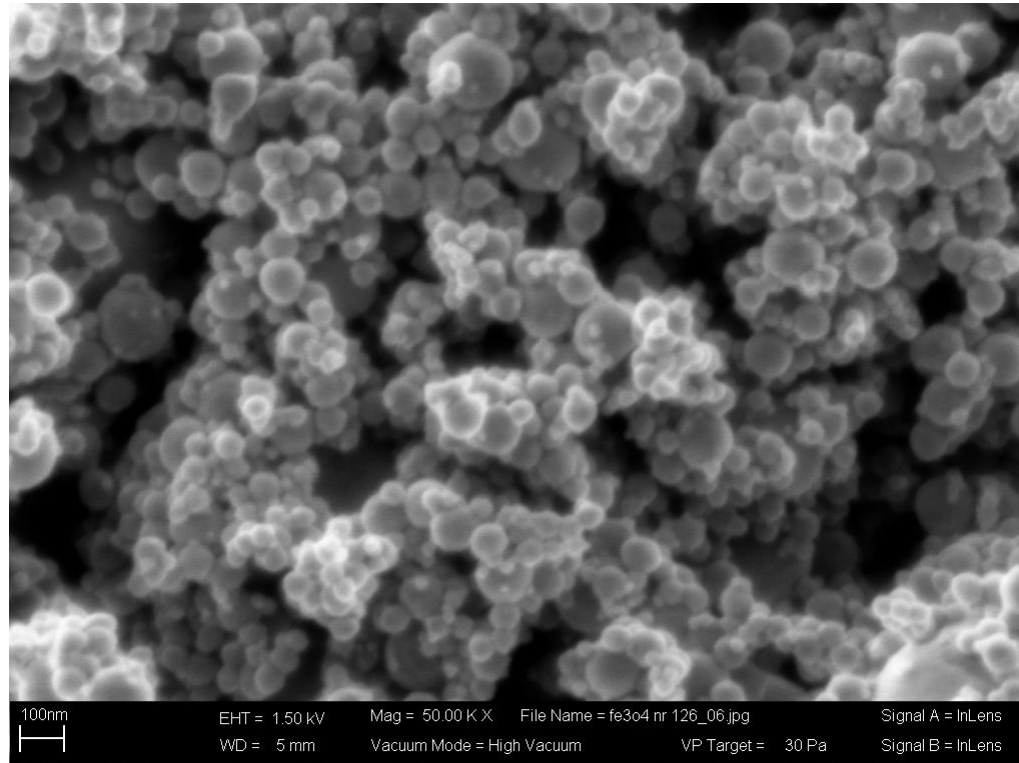
**Figure 9.1** SEM-Record of Nano-Ag-Particles. A stable suspension was applied to an Al-plate and investigated after drying. Particle size approx. 50 nm.





## Example: Ferrimagnetic Nano-Magnetite

11



File: fe3o4 nr 126\_06.jpg

**Figure 11.1** SEM-Record of **Nano-Magnetite-Particles ( $\text{Fe}_3\text{O}_4$ )**. A stable suspension was applied to an Al-plate and investigated after drying. Particle size approx. 50 nm.

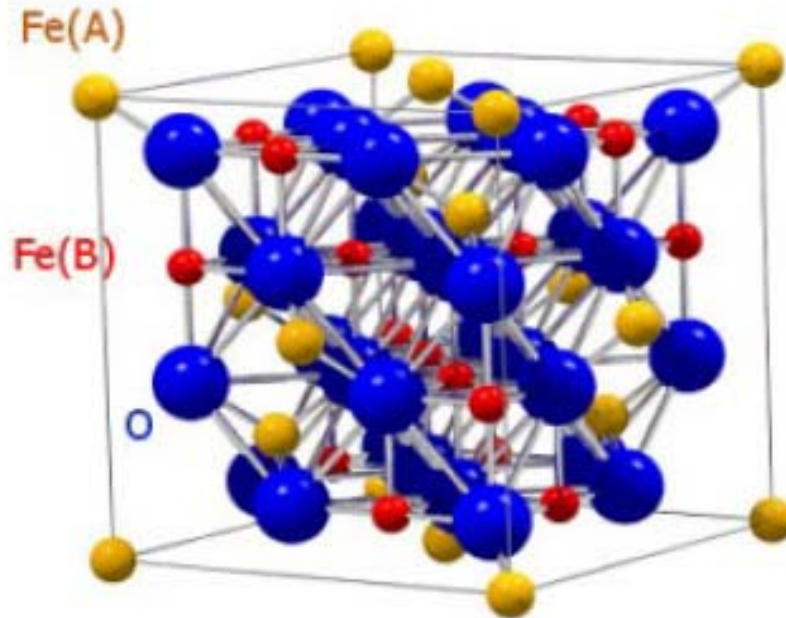


# Crystal Structure of Magnetite $\text{Fe}_3\text{O}_4$

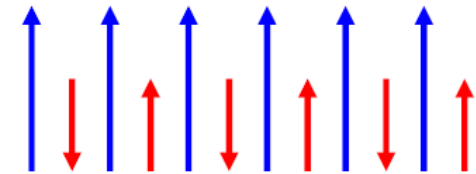
12

+III

+II/+III

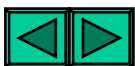


$\text{Fe}^{2+}$  parallel



$\text{Fe}^{3+}$  antiparallel

**Figure 12.1:** Crystal structure of Magnetite. The iron cations are located at two different positions (A and B) of the crystal lattice. The A-position is occupied only by  $\text{Fe}^{3+}$ . The B-position is occupied by  $\text{Fe}^{3+}$  as well as  $\text{Fe}^{2+}$ . The magnetic moments of the  $\text{Fe}^{3+}$  cations in positions A and B are antiparallel and therefore compensate. **Resulting is the magnetic moment of  $\text{Fe}^{2+}$ , which leads to ferrimagnetism.**



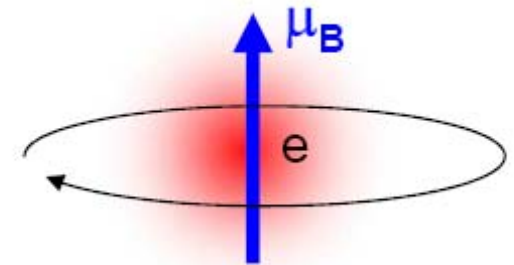
# Example: Ferrimagnetic Nano-Magnetite

13

Atomic number	Element	Electronic structure of 3d	Moment ( $\mu_B$ )
21	Sc		1
22	Ti		2
23	V		3
24	Cr		5
25	Mn		5
26	Fe		4
27	Co		3
28	Ni		2
29	Cu		0

= electronic spin orientation

$$\mu_B = \frac{e\hbar}{2m_e c}$$

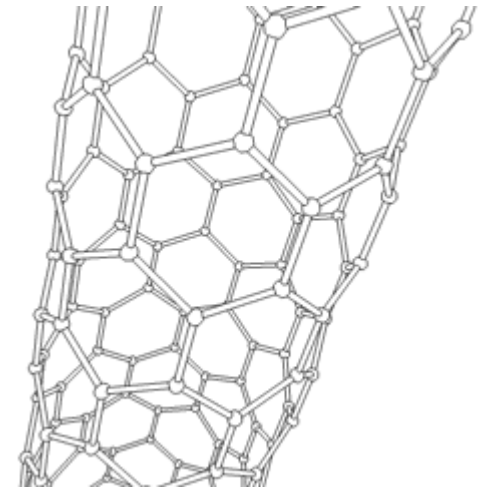
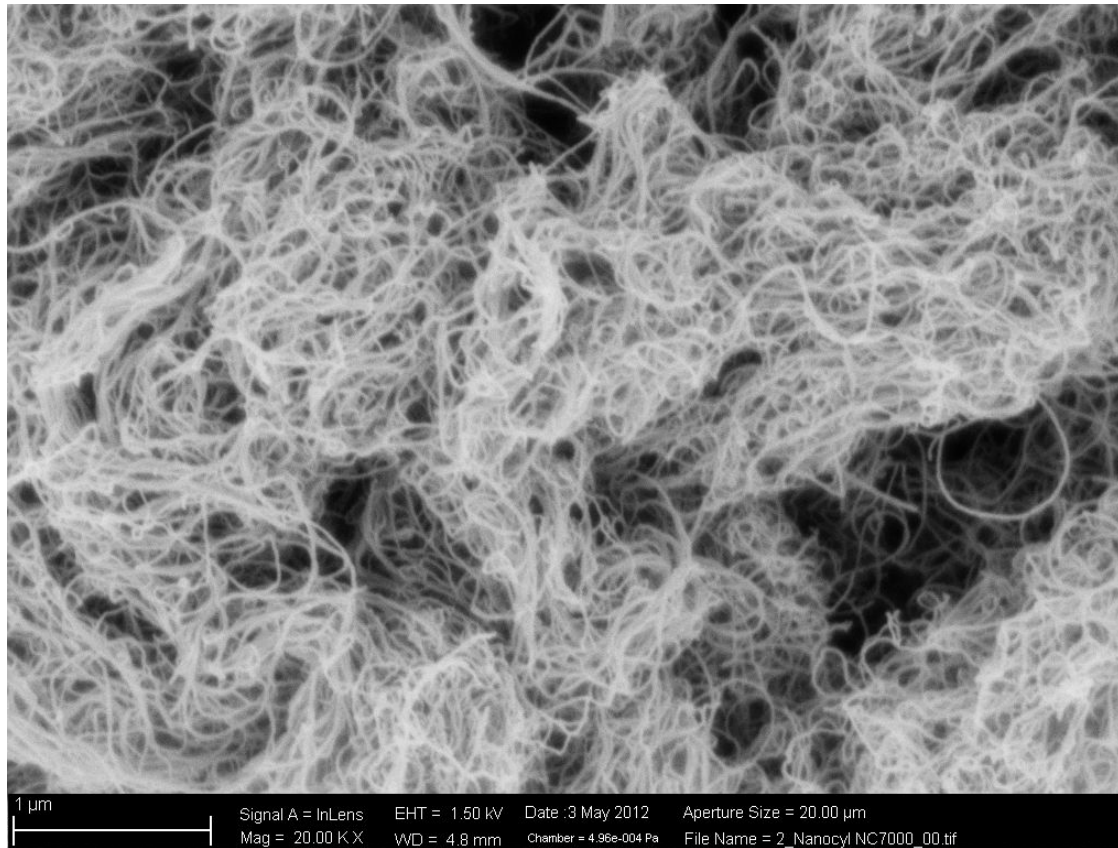




# Example: Degradable Carbo Nano-Tubes for Controlled Drug Delivery

14

Source: Wikipedia

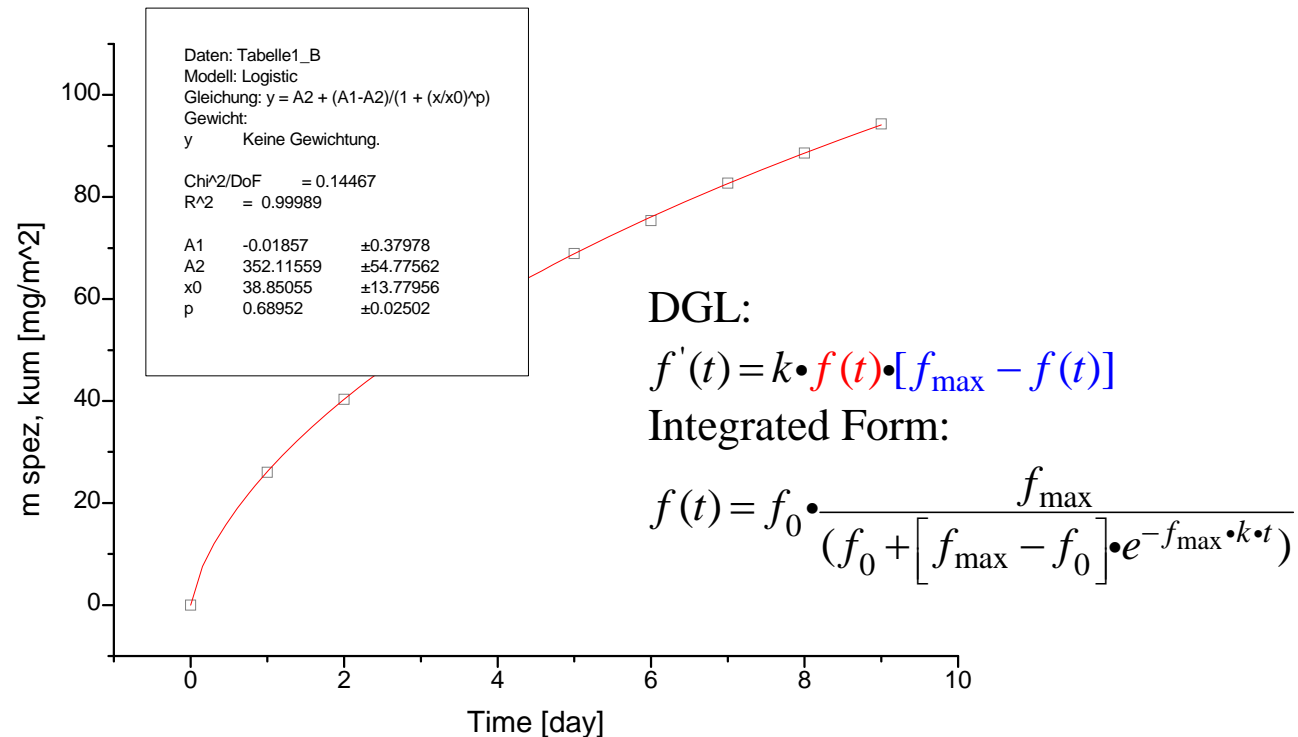


**Figure 14.1** SEM-Record of Carbon Nano-Tubes for Drug Delivery.



# Mathematical Modeling of Cumulative Drug-Release

VModelingBAM.Ppt, 15

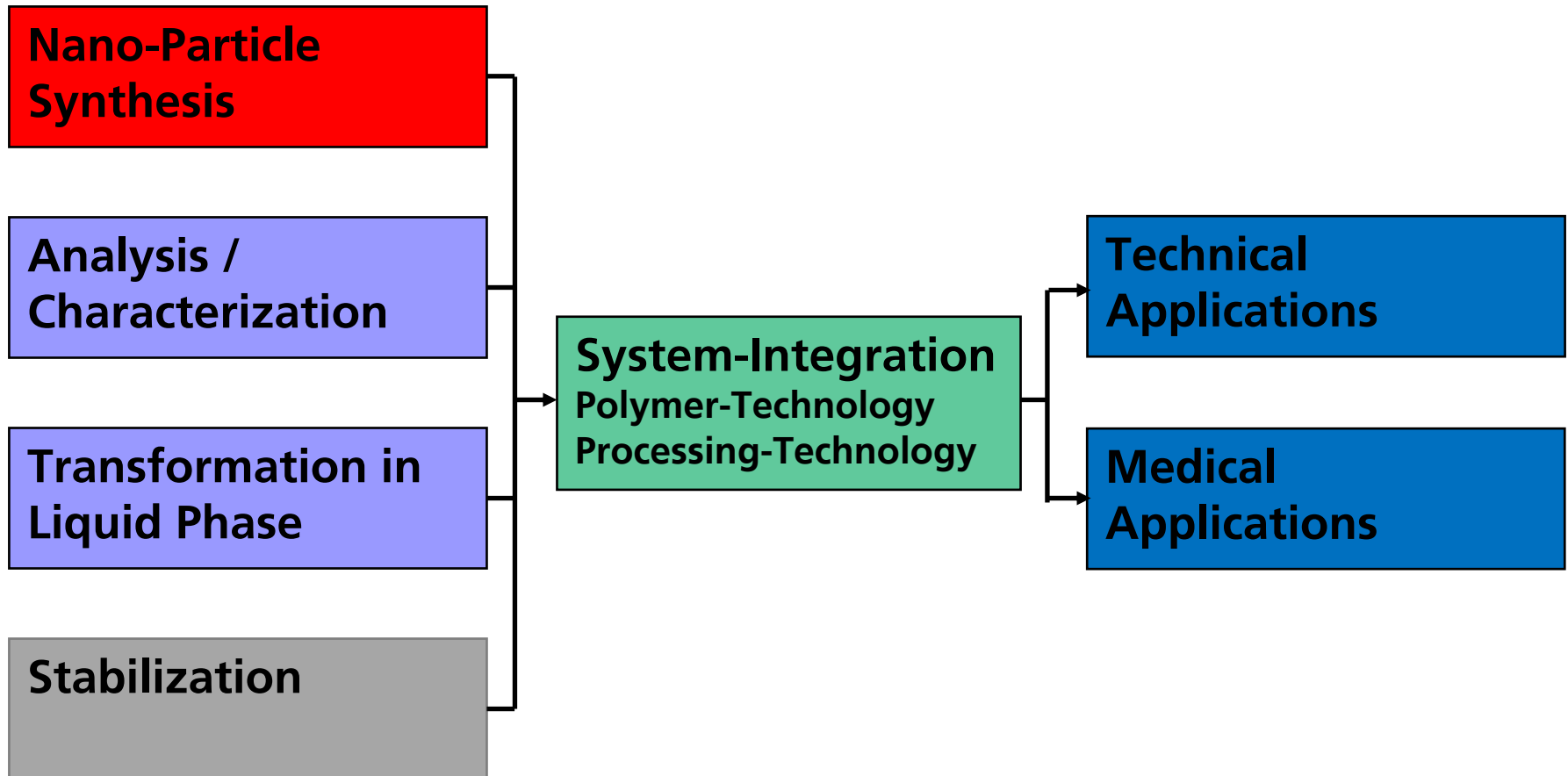


**Figure 15.1** Example for Mathematical Modeling of Drug-Release under Consideration of a Superimposed Solution and Diffusion Process using an Exponential-Function representing „Logistic Growth“.



# Main Steps from Nano-Particle-Production to Applications

16

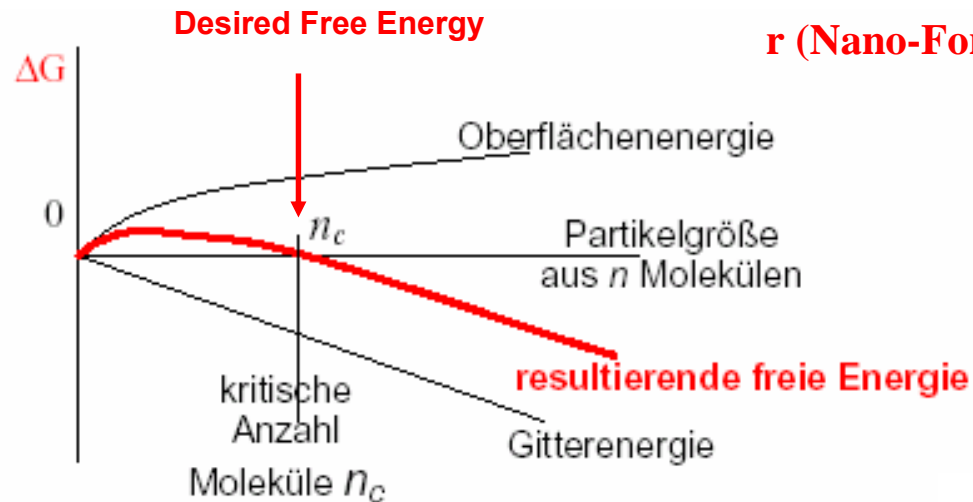


## 2 Methods and General Tools, Nano-Production, Stabilization

17

Source: Penth

### Thermodynamic Effect

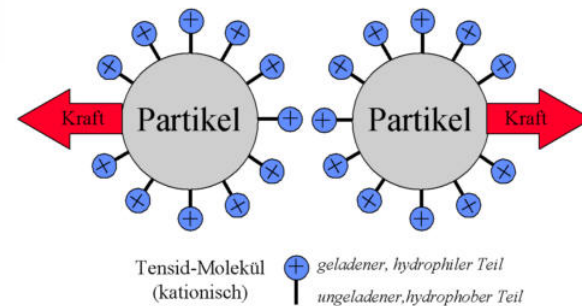


### Kinetic Aspect

$r$  (Nano-Formation)  $\gg$   $r$  (Growth, Agglomeration)

Necessity for Stabilization ->

Surface Treatment



# Theoretical Aspects – Potential and Field-Theory

18

$$\nabla^2 U(\eta, \theta) = \frac{(\cosh \eta - \cos \theta)^2}{B^2} \cdot \frac{\partial^2 U}{\partial \eta^2} + \frac{(\cosh \eta - \cos \theta)^2}{B^2} \cdot \frac{\partial^2 U}{\partial \theta^2} - \frac{\sinh \eta \cdot (\cos \eta - \cos \theta)}{B^2} \cdot \frac{\partial U}{\partial \eta} + \left( \frac{(\cosh \eta - \cos \theta)^2}{\tan \theta} - (\cosh \eta - \cos \theta) \cdot \sin \theta \right) \cdot \frac{1}{B^2} \cdot \frac{\partial U}{\partial \theta} = \sinh(U(\eta, \theta)) \quad (1)$$

U: Reduced Electrostatic Potential, B: Constant in Bispheric System of Coordinates

## Equation 1

Calculation of **surface potential U** according to **Poisson-Boltzmann**-Equation (1) in order to make a suitable selection of chemical additives enabling stabilization and preventing from reagglomeration.

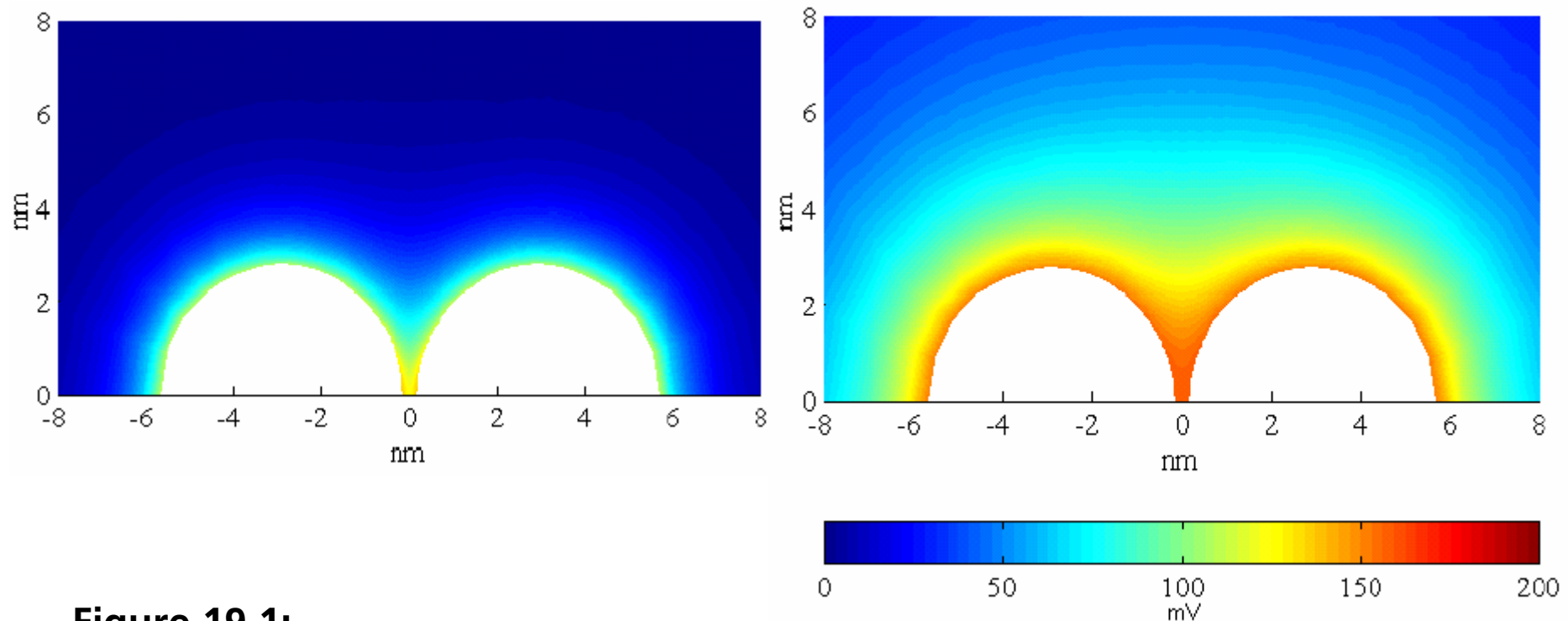




# Theoretical Aspects – Potential and Field-Theory

19

Source: TU-Darmstadt

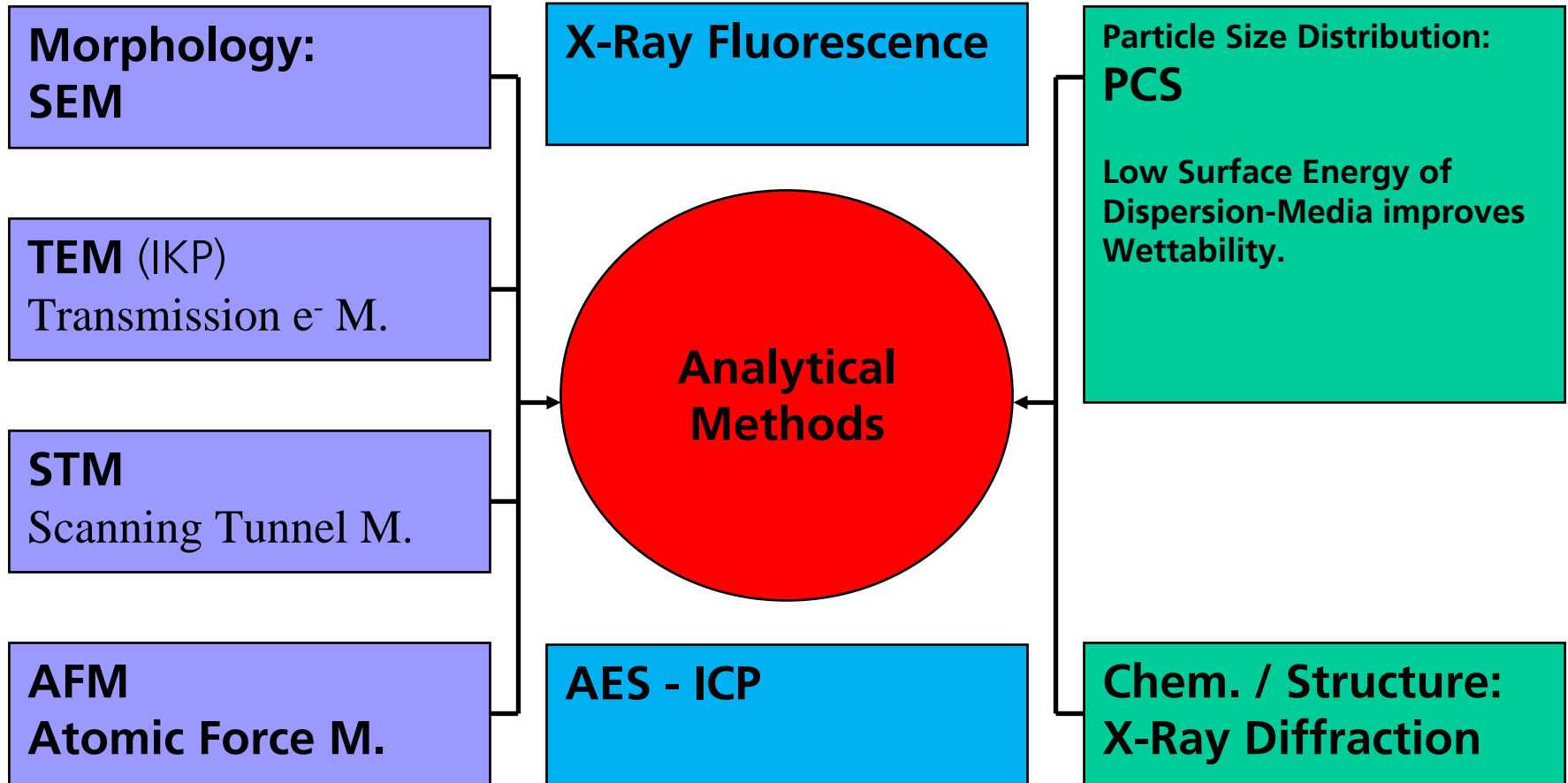


**Figure 19.1:**

Theory of Nano-Particle-Stabilization:

**FE-Calculation of Electrostatic-potential-gradient** between two  $\text{Al}_2\text{O}_3$ -Particles (5 nm) as a Function of pH and Electrolyte-concentration (left pH2, 0.1 M NaCl, right pH4, 0.001 M NaCl)





AAg00103

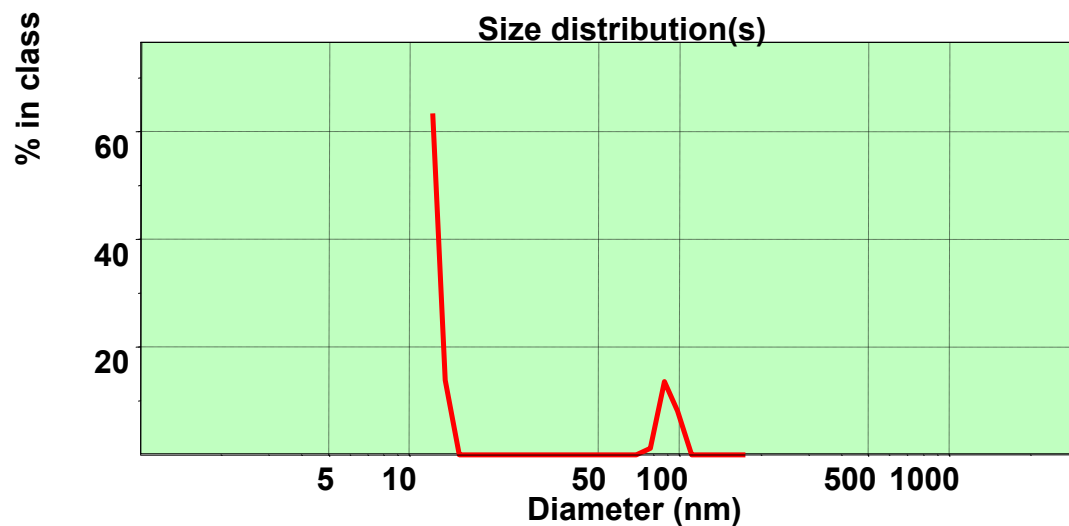
Zetasizer 3000

Data taken on 22/07/04 at 17:07:06

Temperature 25.0 Viscosity 0.890 cP Angle 90.0 deg

RI medium 1.33 RI particle 1.92 + Abs. 1.00

Cumulant Z Ave 46.3 nm Polydispersity 0.599



**Figure 21.1**

**Particle Size Distribution**

Results of **PCS-Analysis**

of Nano-Ag, 1 Vol.-%

Suspension in H<sub>2</sub>O

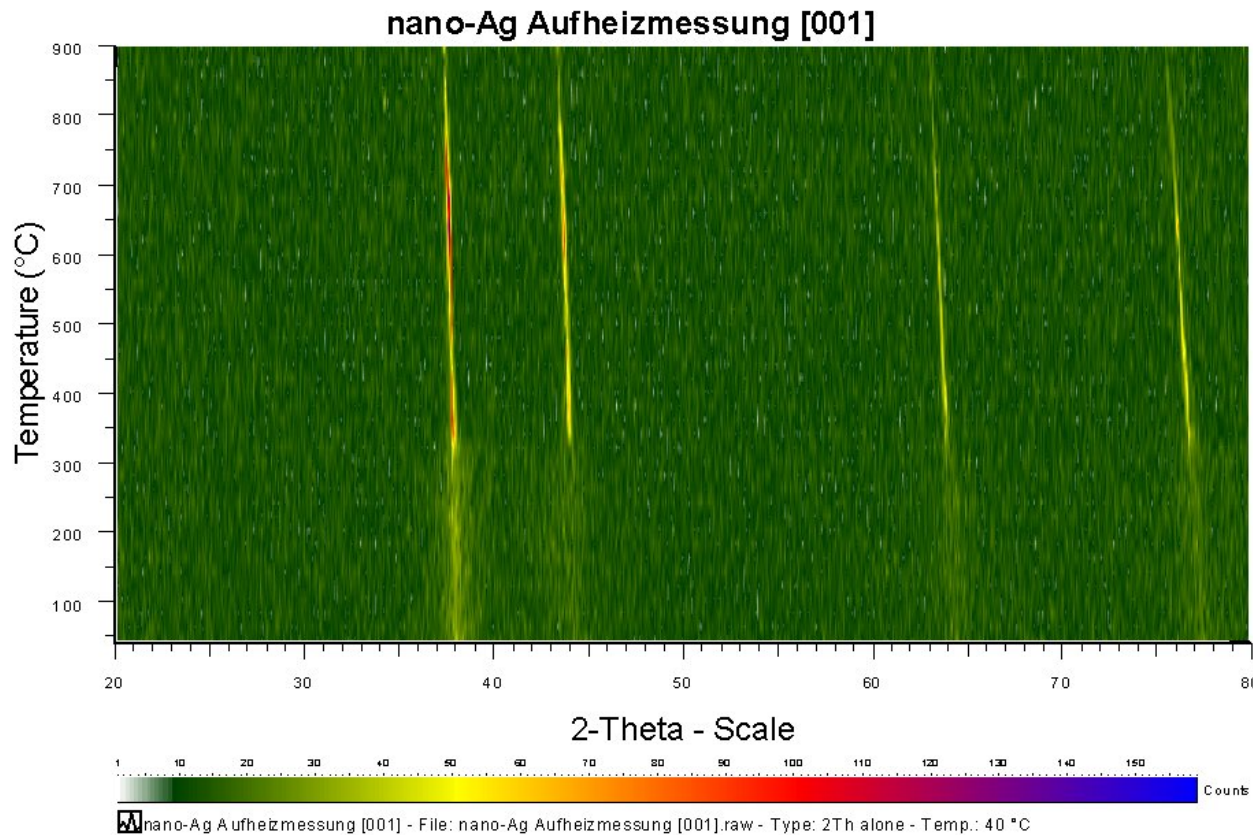
(CAg00103).

Size (nm)	% Intensity	Size (nm)	% Intensity	Size (nm)	% Intensity
12.1	63.3	30.8	0.0	78.2	1.2
13.6	13.8	34.6	0.0	87.8	13.6
15.3	0.0	38.9	0.0	98.7	8.1
17.2	0.0	43.7	0.0	110.9	0.0
19.3	0.0	49.0	0.0	124.6	0.0
21.7	0.0	55.1	0.0	140.0	0.0
24.4	0.0	61.9	0.0	157.3	0.0
27.4	0.0	69.6	0.0	176.8	0.0

Peak : Mean 12.4 width 0.5

# Temperature-resolved X-Ray Diffraction

22

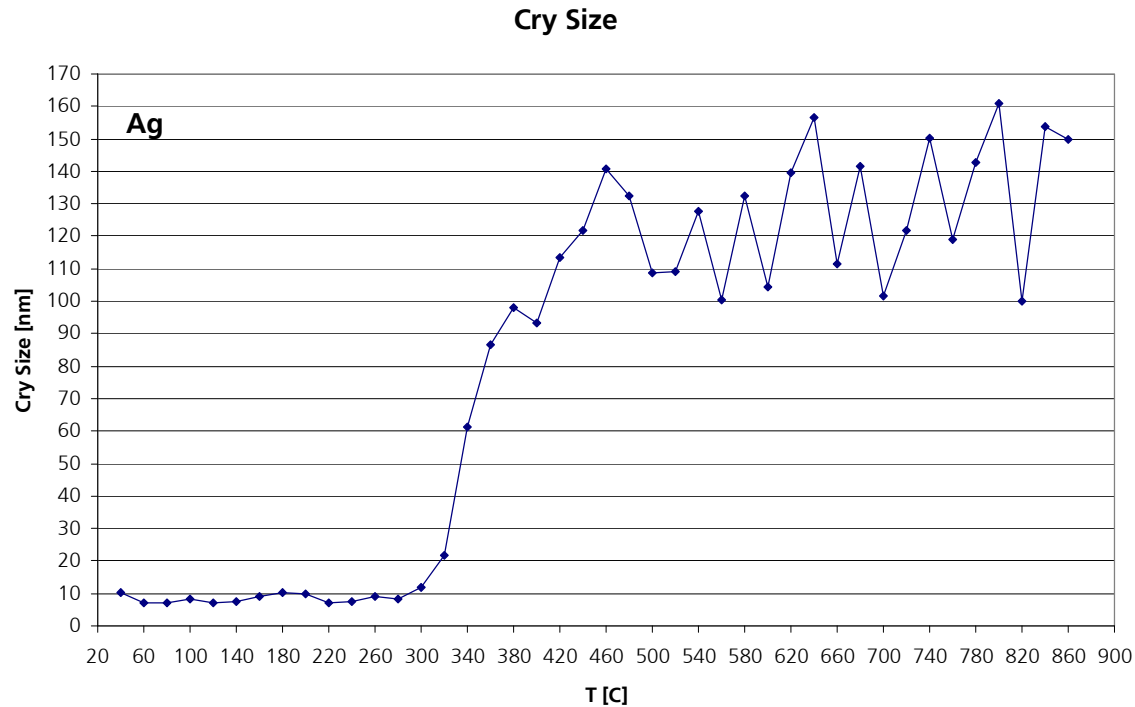


**Figure 22.1** Temperature-resolved **X-Ray Diffraction** Diagrams of a Nano-Ag sample in air in a temperature-range from 40 – 900 °C,  $\chi = 10$  K/min (Fp Ag: 961.9 °C) [Nano-Ag Aufheizmessung\_02.jpg]. Peaks represent Ag-reflexes in different phases.

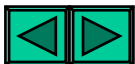


# Temperature-resolved X-Ray Diffraction

23



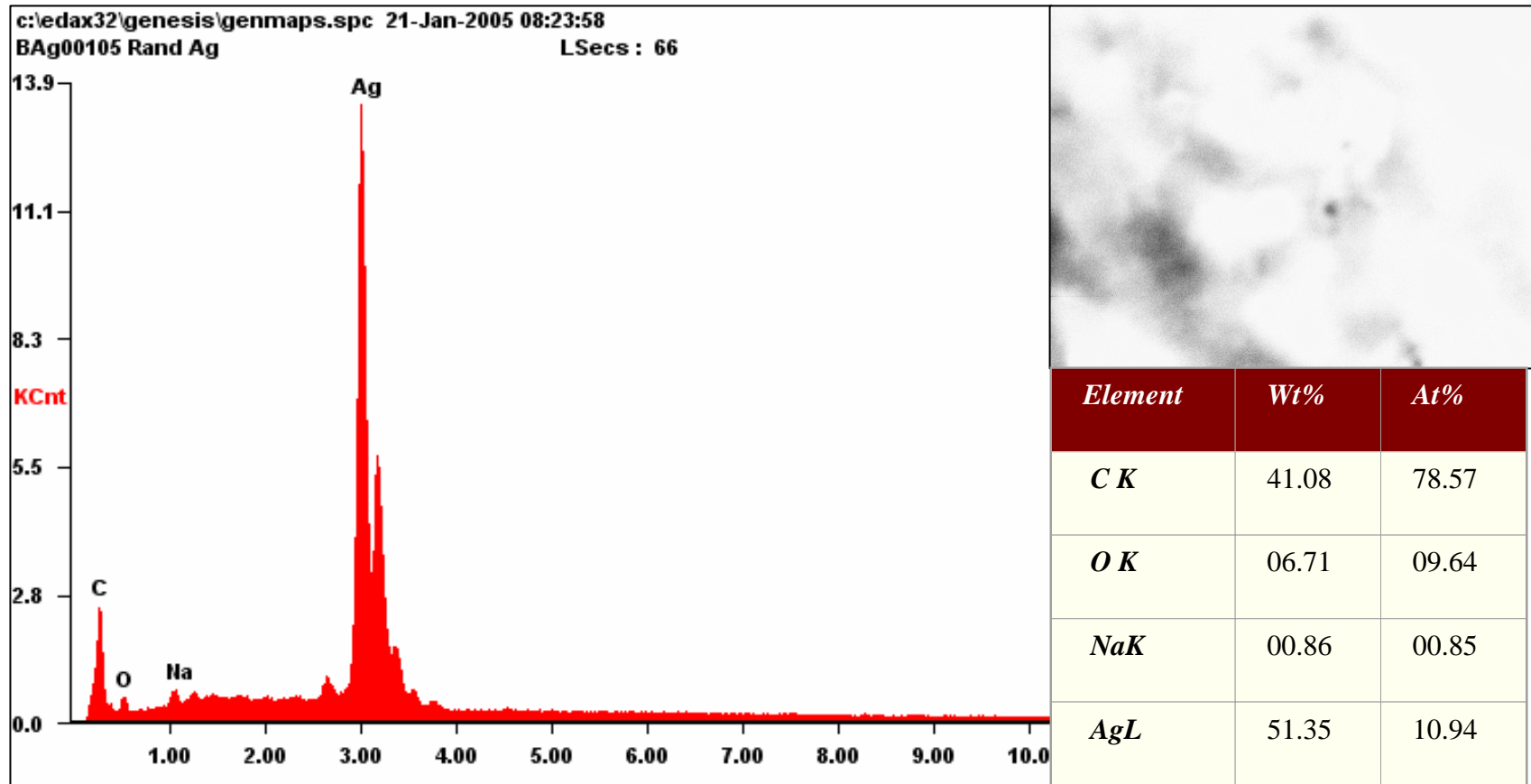
**Figure 23.1** By means of **temperature-resolved** X-Ray Diffraction evaluated primary crystallite dimension of Nano-Ag as a function of temperature. Grain growth starting from 300 °C.





# Energy Dispersive X-Ray Spectroscopy (EDX)

24



**Figure 24.1** Quantitative Ag-Analysis in Sample BAG00105 with X-Ray Fluorescence

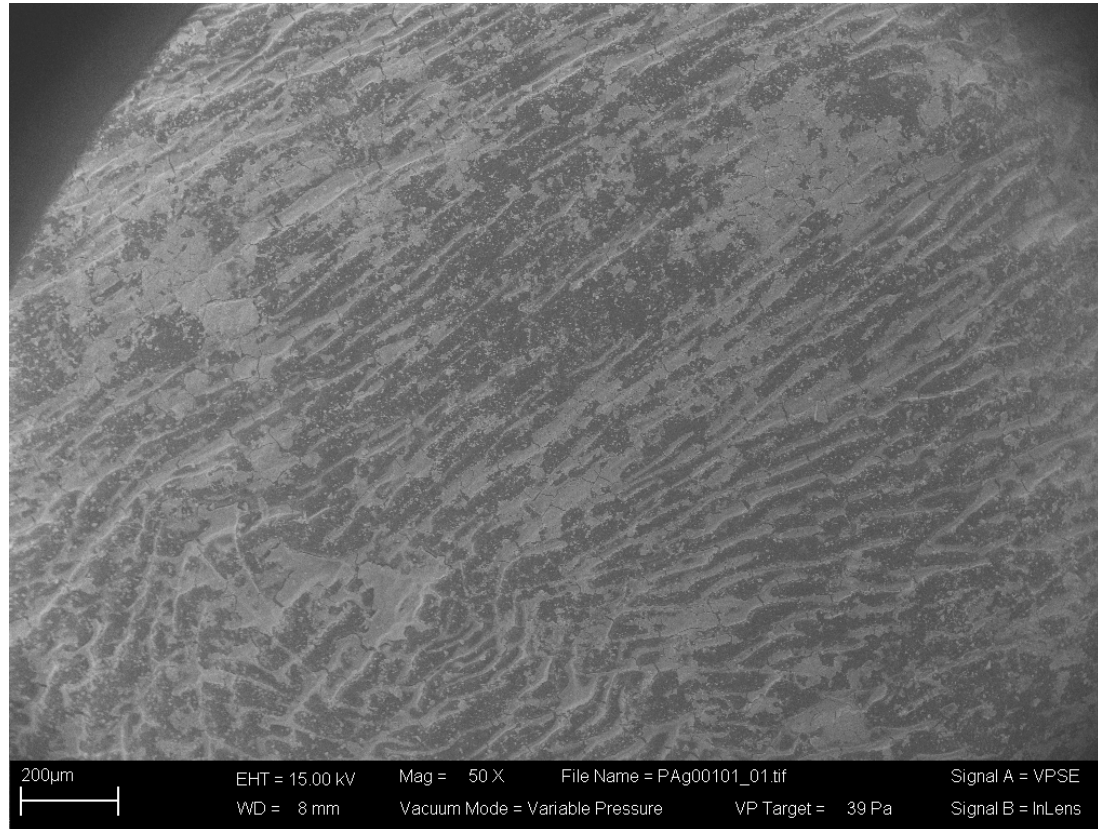




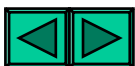
**Figure 25.1** Mini-Twin-screw-extruder for sample-production in larger scale (500 g / h)

# Nano-Strukturing / Coating of Polymer-Granulate to Achieve System Functionality

26



**Figure 26.1** SEM-Data of Nano-Ag-Particles on PP-Granulate (PAg00101).

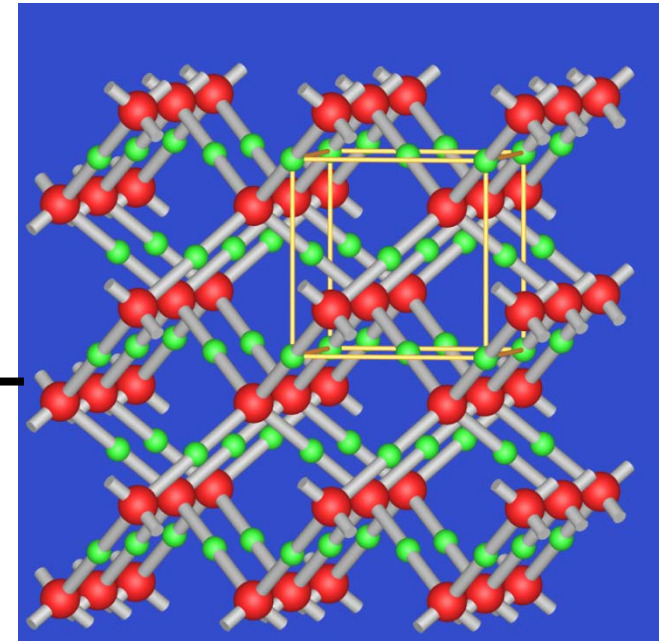
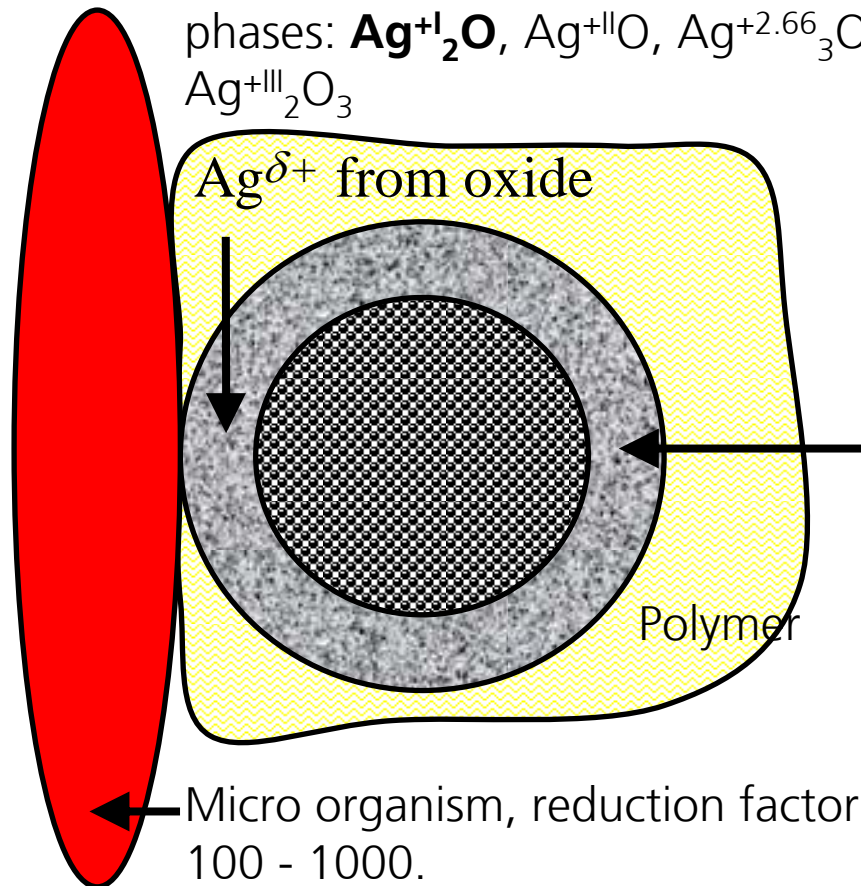


### 3 Results / Discussion, Selected Applications, Nano-Silver-Technology

27

Source: Uni Freiburg

Approx. 25-30 mass-% as porous oxide-phases:  $\text{Ag}^{\text{I}}_2\text{O}$ ,  $\text{Ag}^{\text{II}}\text{O}$ ,  $\text{Ag}^{+2.66}_3\text{O}_4$ ,  $\text{Ag}^{\text{III}}_2\text{O}_3$



Crystal structure  $\text{Ag}_2\text{O}$ :  
Cubic closest packed  $\text{Ag}^+$ ,  
 $\text{O}^{2-}$  occupy  $\frac{1}{4}$  of tetrahedral gaps.

# Ag, Ag<sup>δ+</sup>, Ag<sup>+</sup> as Biocide Substances – Chemical Mechanisms

28

**Electrophile -> Electrophile Mechanism**

**Oxidant, Redox-Mechanism**

**Complex-, Chelate Forming Mechanism**

**Ionic Exchange Reactions**

**Superimposed Processes via Split Ionic Charges**

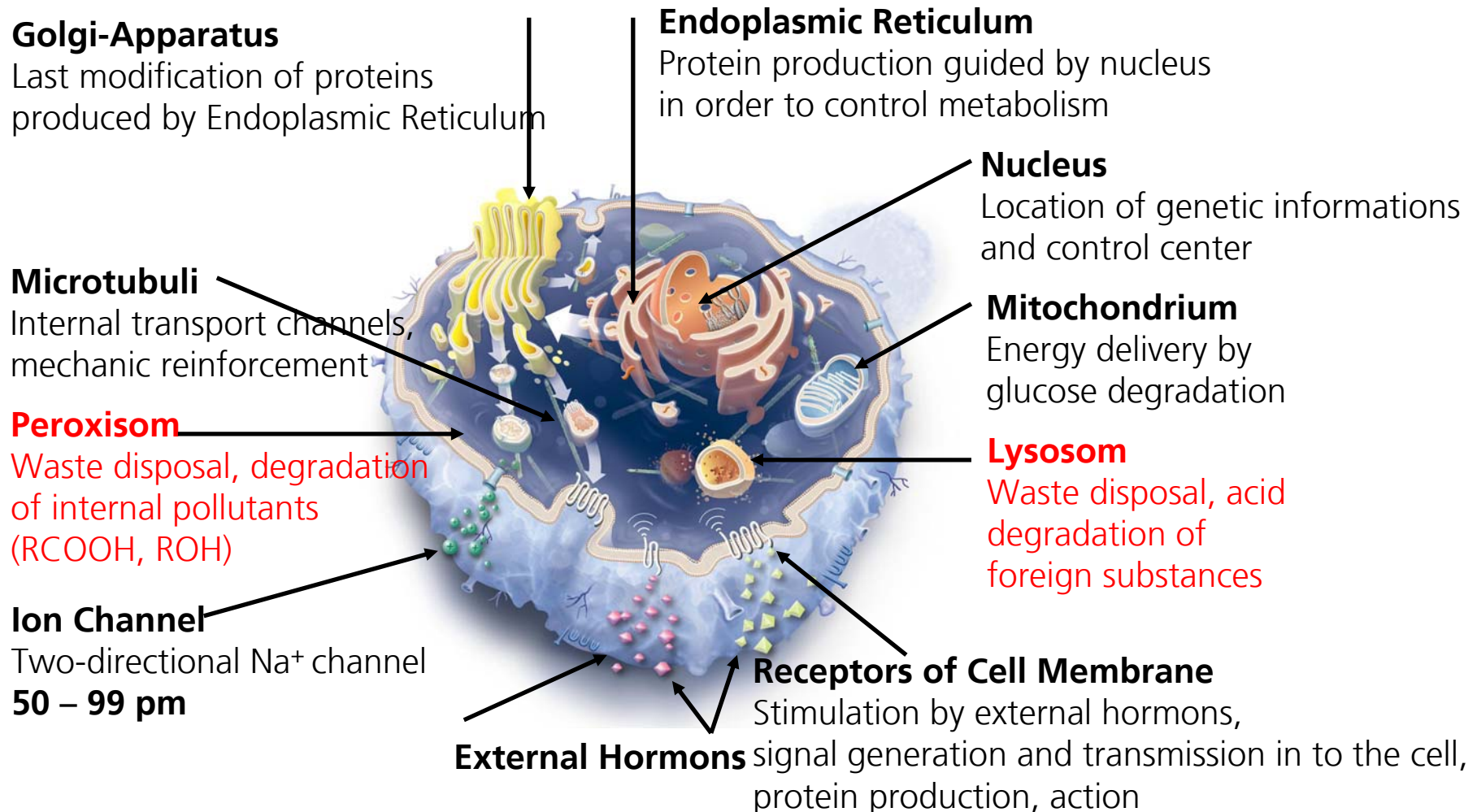




# Biocides, Sub-group Micro-Biocides – Working Mechanism

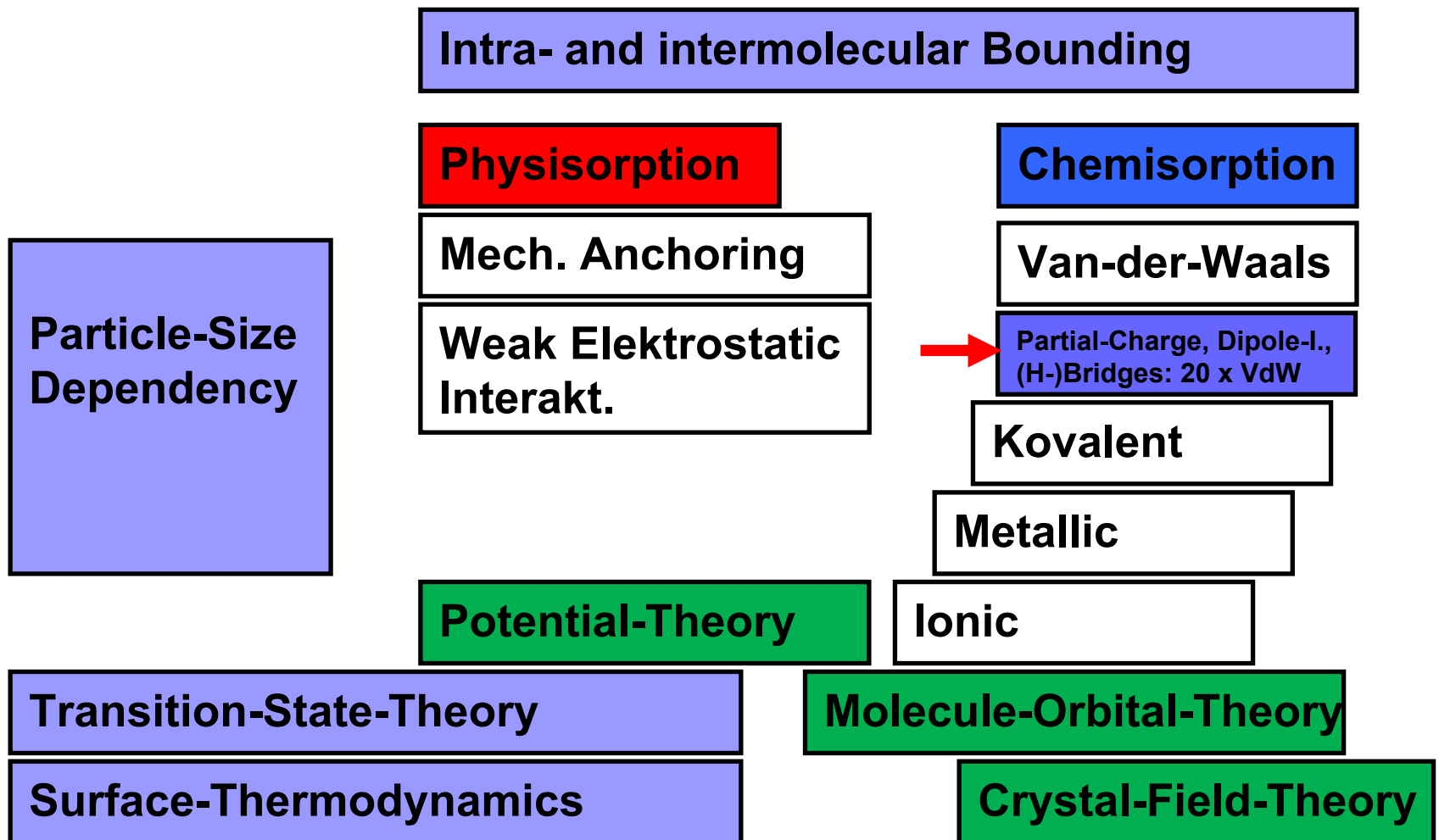
29

Source/Autorization: Wortundbildverlag, 08.02.06



# Theoretical Aspects – Safety Concept

30



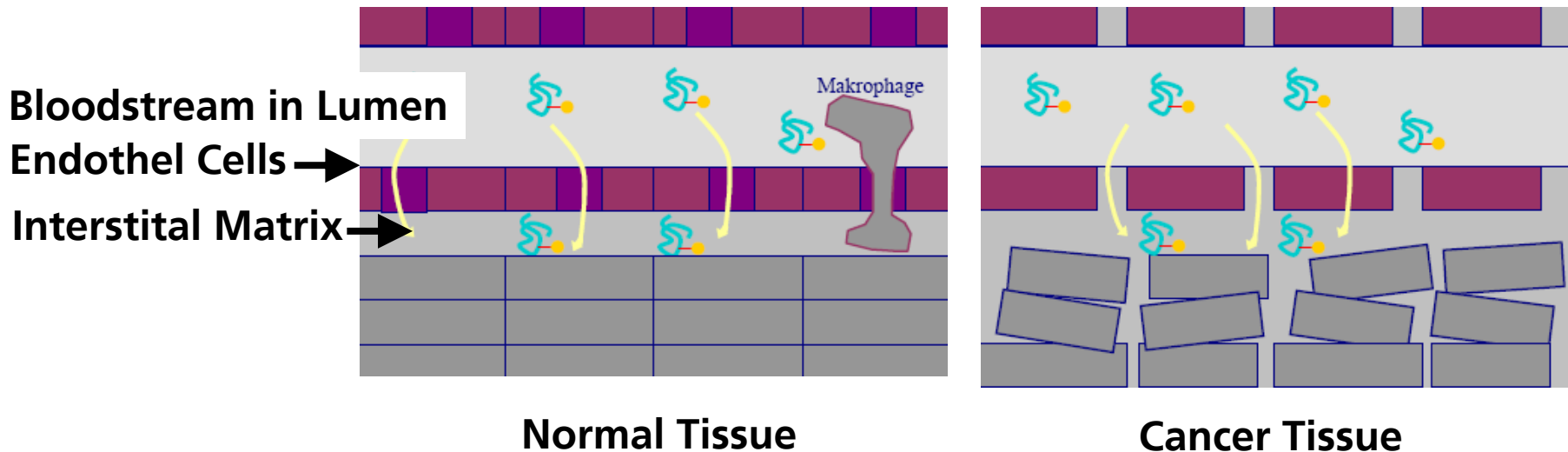
# Transportation, Targeting and Release of Nano-Particles

31

Source: Uni Frankfurt

## Intravascular Transportation

### Passive Targeting -> Accumulation in Reticular Connective Tissue (Liver, Spleen)



## Extravascular, Interstitial Transportation Active Targeting with Antibodies

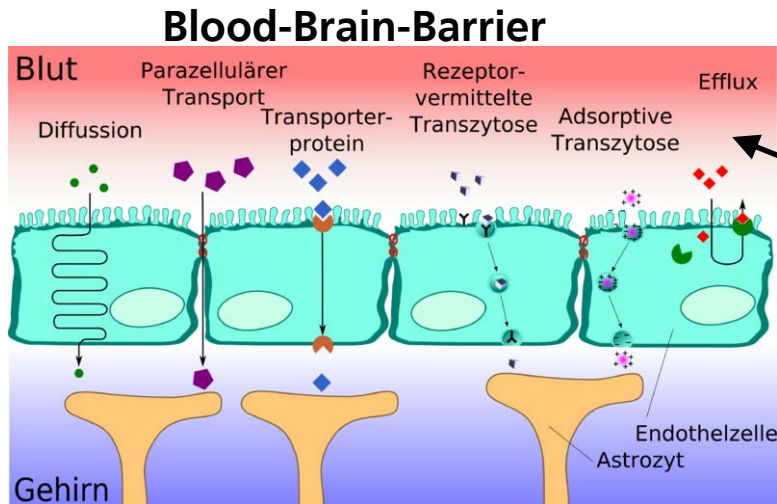


# Capillary Penetration, Blood-Brain-Barrier (BBB)

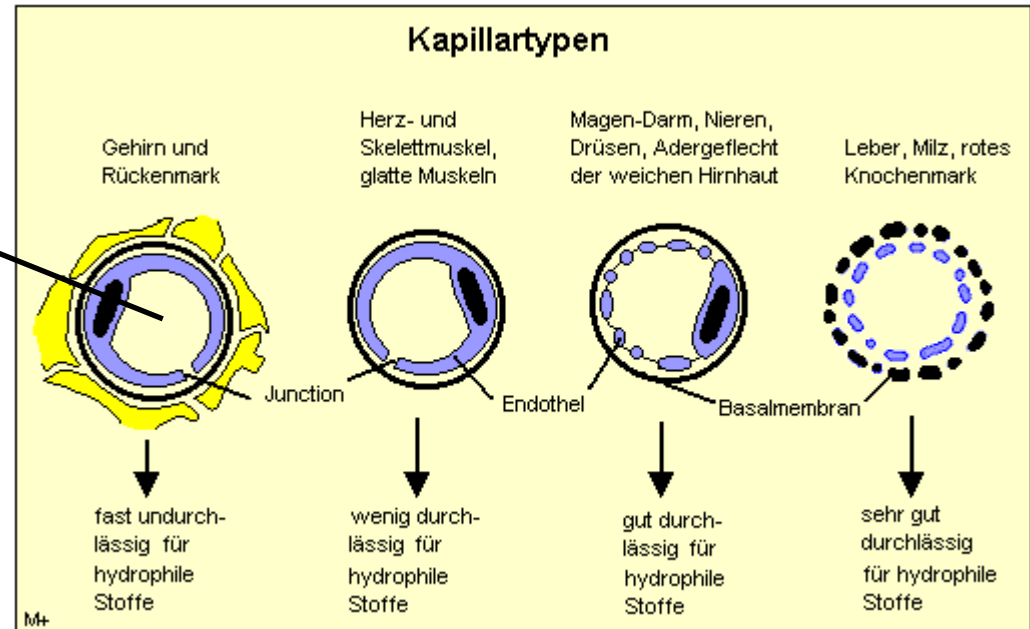
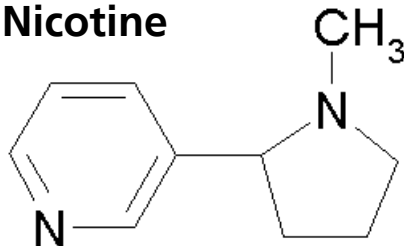
32

Source: wikipedia, medizinfo

Penetration is not only a function of size but of surface charge (hydrophobia, hydrophilicity)



Nicotine



Permeability of hydrophobic substances

Permeability of Hydrophilic substances



# Test-Methods – Fungicide Activity

33

## **Direct-Feed-Test / Direct-Cell-Contakt-Test / Pharm. Eu. 1997**

Sterile delivery.

Sample preparation: Inkubation with 10-100 colony forming units (malt-extract) in a closed test glass, breeding: 3-5 days.

Visual evaluation, comparison with reference.

Additional performance as **Row-Dilution-Test** to evaluate acting concentration.

## **(Konservation-)Stress-Test, similar to Direct-Feed-Test**

No sterile delivery.

Sample preparation as described below.

Visual evaluation.

Additional performance as **Row-Dilution-Test** to evaluate acting concentration.

## **→ Quantitative Suspension-Test**

If necessary, autoclave treatment of sample to prohibit external influence; disadvantage p,T.

0.1 g fungal solution in Mueller-Hinton-Bouillon + 9.9 g sample are to be united.

Germs will be re-isolated.

Rearing / breeding of re-isolated germs.

-Better rearing -> increase of germ number

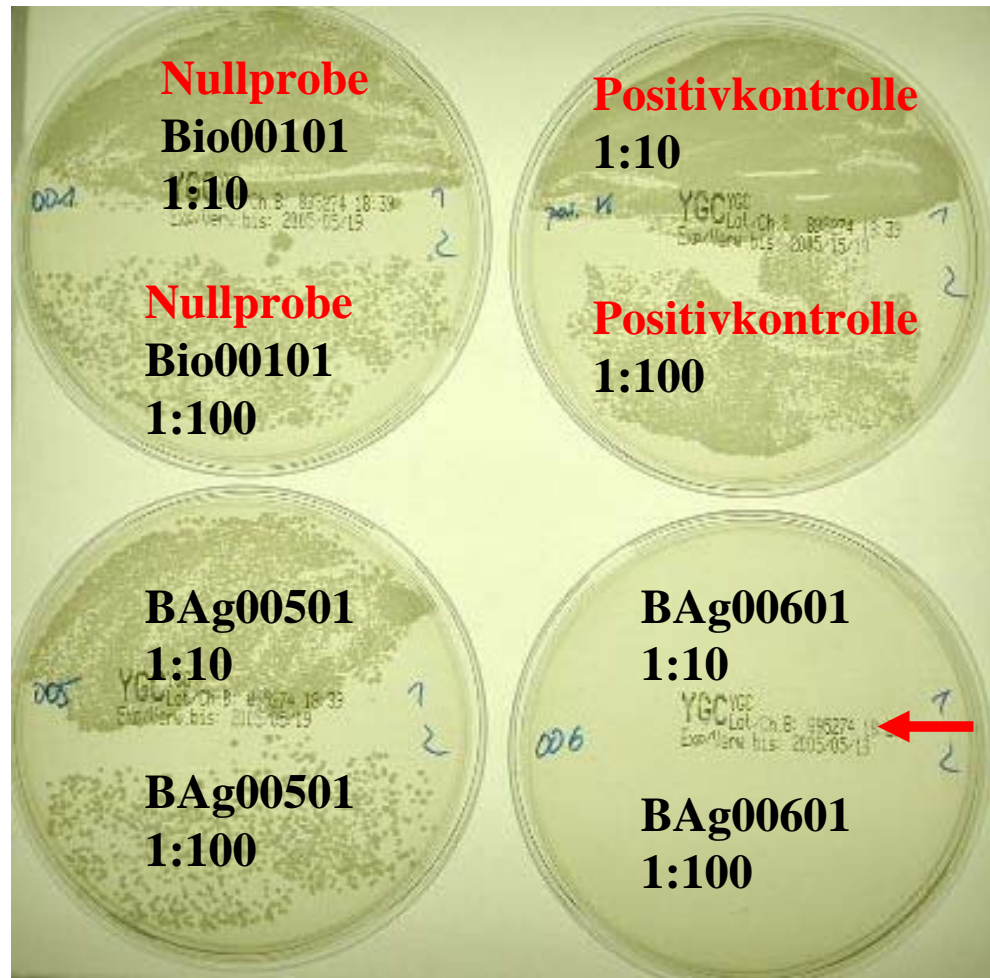
-Normal rearing -> constant germ number

-No rearing -> killing of germs

Additional performance as **Row-Dilution-Test** to evaluate acting concentration.

# Result of Biocide Treatment

34

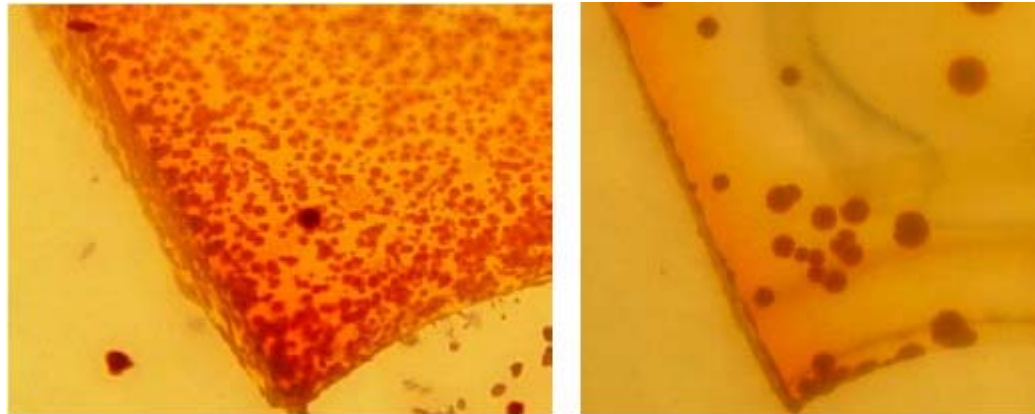


**No fungal attack**

**Figure 34.1** Result of breded diluted crop-out of the samples (mP/mNL=const.)

# Antimicrobial Coatings

35



**Figure 35.1** Example of a Nano-Ag-coated Silicon-Polymer (on the right) compared to a untreated sample (on the left) with colonies of **Staphylococcus Aureus**.

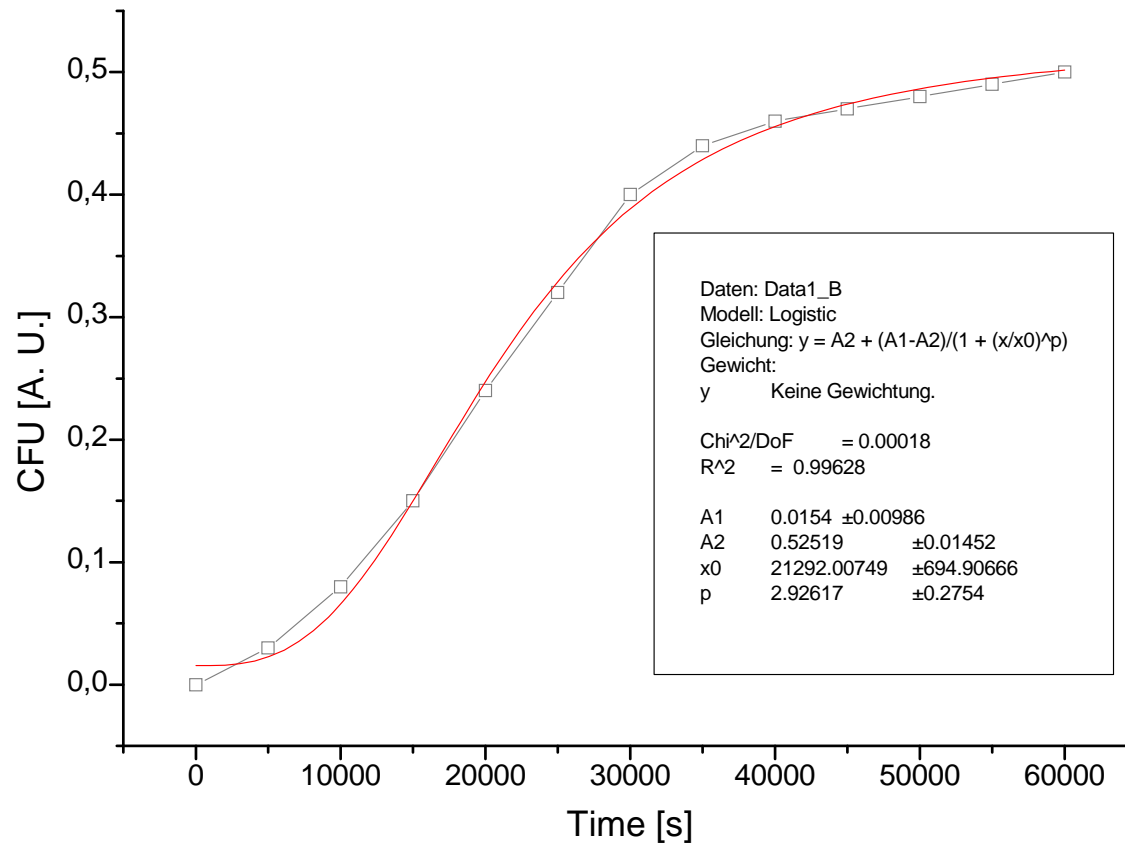
A reduction rate of factor 100 was detected based on the applied Nano-Ag-content (Test-setup: Spray-test).

**Depending on Nano-Ag-content reduction rates of 99,999 % (5 log-levels) are possible.**



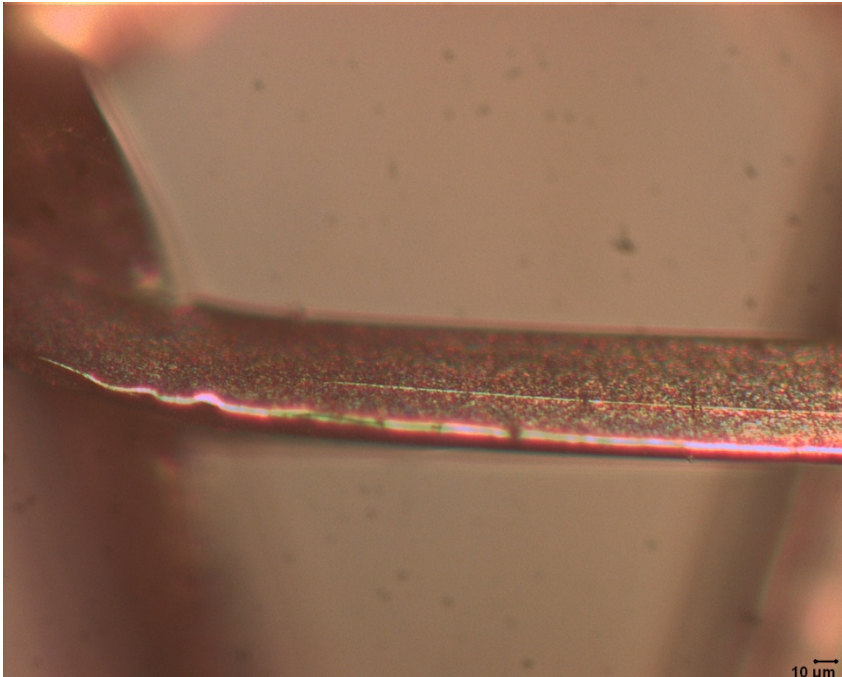
# Mathematical Modeling of Microbial Growth

36



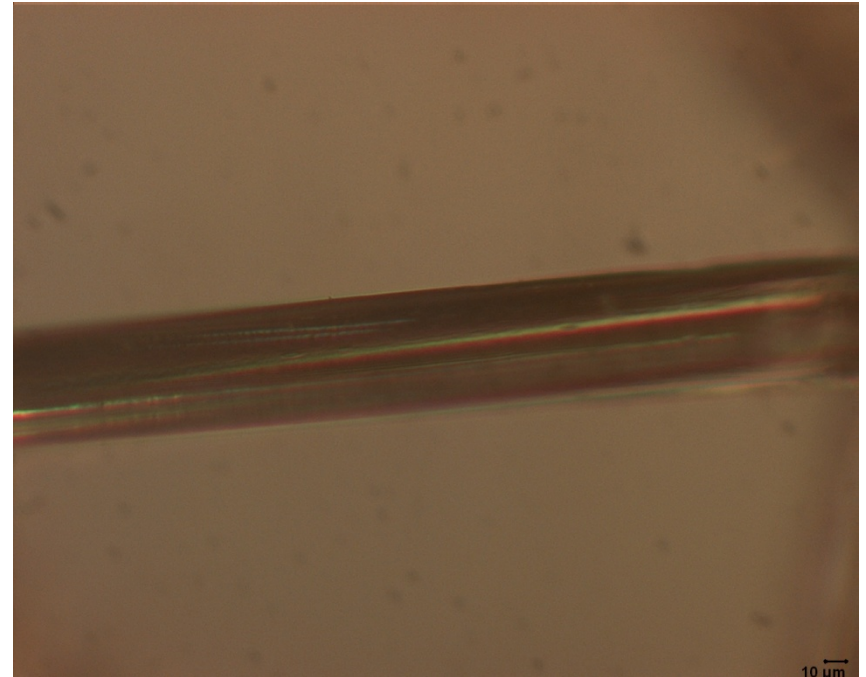
**Figure 36.1** Example for Mathematical Modeling of **Microbial Growth** (Exp. Data) with an Exponential-Function representing „Logistic Growth“.





**Figure 37.1**

Microscopic image (incident-microscopy) of a **metal-coated PA-fibre** (File: TüllA.Jpg, 33%)

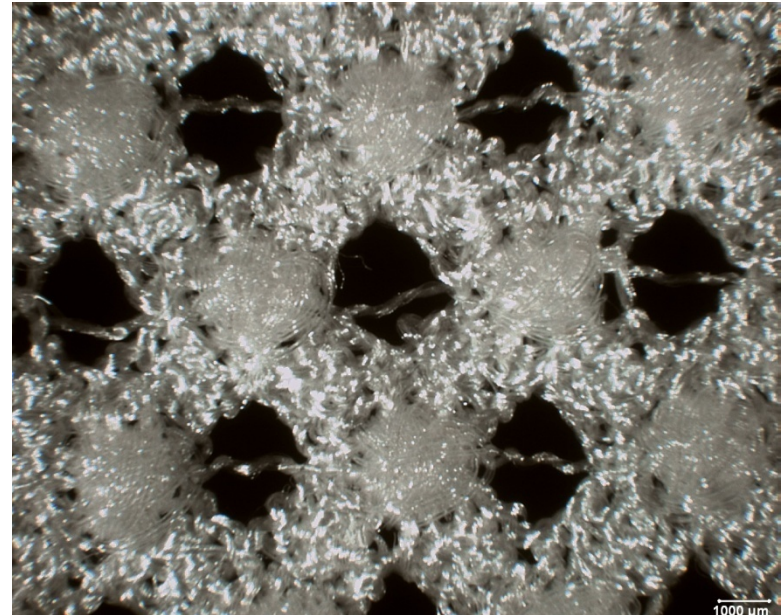
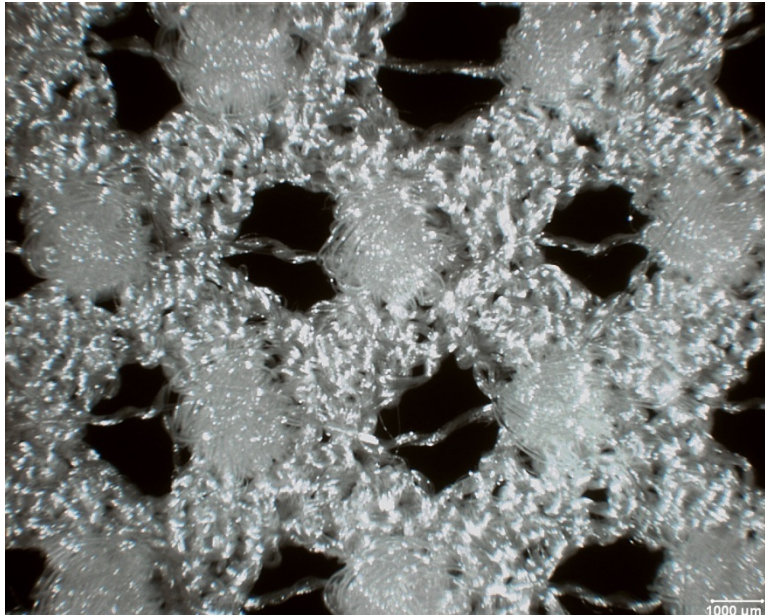


**Figure 37.2**

Microscopic image (incident-microscopy) of a **nano-metal-coated PA-fibre** (File: TüllRohA.Jpg, 33%)

# Wound-Pad Application

Msoffice\Powerpoint\Nano.Pnt\VNano.Ppt, 38



**Figure 38.1** Macroscopic Image of a PE-Biopad NURPES-001 (Wound-Pad). Left sample uncoated, right sample Nano-Ag-coated,  $x < 100$  ppm (GAg00001, GAg00101).

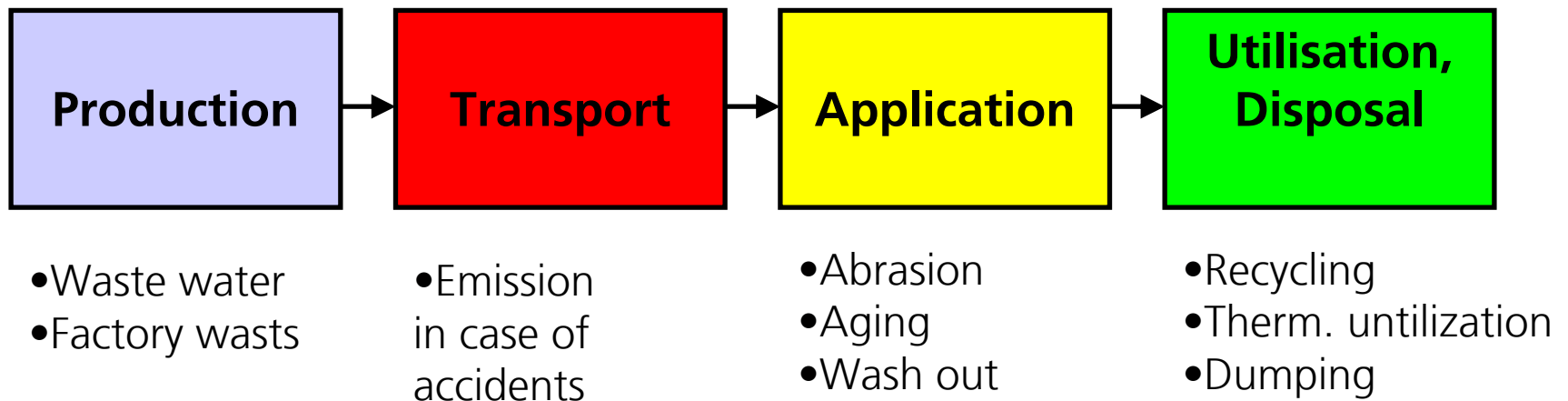
Goal: Medical Device Class 2b.



# Product Example: Antimicrobial Nano-Silver-Dispersion-Paint

39

## Product Safety by Life Cycle Analysis



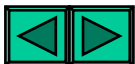


# Product Example: Antimicrobial Nano-Silver-Dispersion-Paint

40



**Figure 40.1** Al Wasl Hospital, **Dubai**, U.A.E.



# Product Example: Antimicrobial Nano-Silver-Dispersion-Paint

41



**Figure 41.1** Al Wasl Hospital (Building 2), Dubai, U.A.E.





# Product Example: Antimicrobial Nano-Silver-Dispersion-Paint

42



**Figure 42.1** Hospital in **Moscow** which was equipped with the developed Nano-Silver-Dispersion-Color as a protection against Staphylococcus Aureus (MRSA).





---

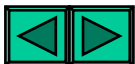
## 4 Abstract / Summary

---

43

Nanotechnology Special Branch, Product Area Energetic Systems at ICT has it's key competences according to following items:

1. **Production** of nearly all kinds of Nano-Particles (metals, transition-elements, oxides) on laboratory scale as well as in bulk quantities
2. Adaptation process of **particle size** distribution developed
3. Production of **stabilized** Nano-Suspensions -> Interface to application
4. Know-how of **system integration** especially in combination with polymer systems
5. **Nano-Products** already established with several years of field experience
6. Nano-**Safety-Concept** developed and analytically proofed
7. Full **compatibility to all legislative injunctions** / homologations
8. Until now the developments were recognized with 5 national resp. international **awards**



## 5 Prospective

---

44

- **Further Nano-Products with additional antimicrobial effect, will be put to market soon**
- **Numerous promising systems are still under development**
- **Nanomedicine is identified as an outstanding field of application**
- **Recognizing the overall benefit, investments in this technology makes sense. Fraunhofer-ICT is looking for cooperations in order to perform in-vitro and in-vivo experiments**



**Dziekuje za uwage!**

**Danke für Ihre Aufmerksamkeit!**

**Thanks for your attention!**

